

Communication

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Asymmetric Addition of Alkylzinc Reagents to Cyclic α , β -Unsaturated Ketones and a Tandem Enantioselective Addition/Diastereoselective Epoxidation with Dioxygen

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We recently reported the asymmetric addition of alkylzinc reagents to ketones using an easily prepared constrained-geometry catalyst.¹ A noteworthy aspect of this study is that it offers a solution to a long-standing problem in synthetic organic chemistry: the catalytic synthesis of tertiary alcohols from acetophenone derivatives with high enantioselectivity. In contrast to related addition reactions with aldehyde substrates, which are promoted by hundreds of catalysts,^{2,3} only a handful of systems will promote additions to ketones,^{1,4-7} and most of these require high catalyst loadings and long reaction times.^{5–7} Despite significant effort in this area, we are not aware of successful reports of the enantioselective addition of alkyl groups to conjugated cyclic enones. In this Communication, we report the results of an investigation into the efficient and highly enantioselective 1,2-addition reactions of cyclic α,β -unsaturated ketones. The resulting tertiary allylic alcohols are valuable intermediates that can be further elaborated via directed epoxidation reactions with excellent diastereoselectivities. The resulting epoxy alcohols undergo Lewis acid promoted semipinacol rearrangement to give β -hydroxy- α , α -disubstituted ketones. Furthermore, we have developed a tandem, one-pot protocol for the enantioselective addition/diastereoselective epoxidation simply by capping the reaction with a balloon of dioxygen when the asymmetric addition is complete.



Initial evaluation of ligand 1 in the asymmetric addition reaction (eq 1) with ZnMe₂ and ZnEt₂ indicated that 2-substituted enones are excellent substrates. Thus, as shown in Table 1, treatment of enone 2 with 10 mol % ligand 1, 3 equiv of ZnMe₂, and 1.2 equiv of Ti(OⁱPr)₄ at room temperature leads to the tertiary allylic alcohol in 84% yield and 99% enantioselectivity after chromatography on silica gel. Likewise, use of ZnEt2 under identical conditions provided the allylic alcohol in 76% yield and 98% enantioselectivity (entry 2). It is significant that the titanium catalyst is chemoselective, giving only carbonyl addition and no conjugate addition. In contrast, copper-catalyzed additions of dialkylzinc reagents to enones give conjugate addition products.8 In a related titanium-BINOL-catalyzed

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Table 1	Asymmetric Addition of Alkyl Groups to Cyclic Enones	

entry	substrates		ZnR_2	time (h)	yield (%)	ee ^a (%)
1	Ŷ		R = Me	40	84	99
2		2	R = Et	40	76	98
3	Ĵ,	3	R = Me	40	55	98
4	$\langle \gamma \rangle$	-	R = Et	37	65	96
5	C ₅ H ₁₁		R = Me	40	62	99
6		4	R = Et	40	50	99
7	7 0 Ph 8 0 Ph		R = Me	40	54	95
8		5	R = Et	40	40	95
9			R = Me	38	20	99
10	Of Ph	6	R = Et	38	32	99
11		7	R = Et	19	75	52
12		8	R = Et	80	50	61

^a See Supporting Information for ee determination protocol.

asymmetric addition of alkylzinc reagents to aldehydes, it has recently been definitively demonstrated that the alkyl group is transferred from titanium, and not zinc.9 Thus, the oxophilic Tialkyl adds in a 1,2-fashion, and the soft Cu-alkyl adds in a 1,4fashion.

The scope of the asymmetric ketone alkylation is illustrated in Table 1. Addition of ZnMe₂ and ZnEt₂ to 2-methyl cyclopentenone exhibits 98% and 96% enantioselectivity in 55% and 65% yield, respectively. Increasing the length of the 2-substituent to pentyl led to an increase in the enantioselectivities to 99% for both ZnMe₂ and ZnEt₂. Aryl-substituted substrates also underwent alkylation with high enantioselectivity, but with diminished yields due to aldol/ dehydration side reactions.

Reactions of 2-phenyl cyclohexenone with ZnMe2 and ZnEt2 both generated allylic alcohol products with 95% enantioselectivity, albeit in 54% and 40% yield. Decreased yields were observed with the exocyclic enone 6, which gave 20% and 32% yield with 99% enantioselectivity in each case. Enones devoid of alkyl substituents α to the carbonyl group are not particularly good substrates, giving 52% and 61% enantioselectivity (entries 11 and 12). Because of

Table 2. Semipinacol Rearrangement	it Products
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ent	ry substrates	products	yield (%)	ee (%)
1	Me OH C ₅ H ₁₁	Me 	70	99
2	Et OH C ₅ H ₁₁		89	99
3	Me Me We		75	98
4	Et OH Me	Et Me	73	97

the similar steric environments of the oxygen lone pairs, discrimination of the enantiotopic carbonyl faces of these substrates is quite challenging.

The allylic alcohols in eq 1 are valuable chiral building blocks that contain chiral quaternary centers. Such centers are difficult to access, and, therefore, methods to synthesize them have attracted considerable attention.^{10,11} The Lewis acid-promoted semipinacol rearrangement of α -hydroxy epoxides is another useful method to establish chiral quaternary centers.^{12,13} Thus, diastereoselective epoxidation of selected allylic alcohols was examined using *m*-CPBA. The desired *syn*-epoxy alcohols were obtained in >90%yield. Treatment of the epoxy alcohols with BF3. OEt2, as shown in eq 2, resulted in semipinacol rearrangement cleanly providing new β -hydroxy- α , α -disubstituted ketones in good yields (Table 2). The ketone products are formed as single diastereomers with no loss of ee in the epoxidation/rearrangement sequence. These densely functionalized products contain chiral all-carbon stereocenters (entries 1, 2, and 4).

We next investigated the possibility of performing a tandem onepot enantioselective ketone alkylation/diastereoselective epoxidation sequence. Dialkylzinc reagents readily react with dioxygen to provide metal peroxides.^{14,15} In the presence of titanium tetraisopropoxide, hydroperoxides are known to catalyze olefin epoxidation with high diastereoselectivity.¹⁶ Performing the addition reaction with ZnEt₂ followed by exposure of the reaction mixture to dioxygen (1 atm) resulted in complete conversion to the syn-epoxy alcohol in under 4 h. Epoxidations in the presence of ZnMe₂ were very slow. In these reactions, 2 equiv of ZnEt₂ was added before exposure to dioxygen. The yields and enantioselectivities of the epoxy alcohols (Table 3) are essentially identical to those of the addition reactions reported in Table 1. Interestingly, the epoxidation reaction outlined here is very different than the Sharpless asymmetric epoxidation in that our reaction is conducted under basic, aprotic conditions.¹⁷

An effective protocol for the enantioselective addition of alkyl groups to cyclic conjugated enones is disclosed. This transformation



Table 3. One-Pot Synthesis of Epoxy Alcohol Products with Oxygen

entry	substrates	ZnR_2^a	yield (%)	ee (%)
	0			
1	Ĭ.	R = Me	82	99
2	2	R = Et	80	99
3	Ŷ.	R = Me	50	98
4	3	R = Et	60	97
5		R = Me	67	99
6	4	R = Et	60	99
7	Ph 6	R = Et	34	96

^a For reactions with ZnMe₂, 1.5-2 equiv of ZnEt₂ was added after complete ketone consumption.

can be coupled with a diastereoselective epoxidation in which the ultimate oxidant is generated from dioxygen. Rearrangement of the resulting epoxy alcohols provides highly functionalized aldol type products with excellent ee that are not accessible through traditional aldol chemistry. We are currently investigating the mechanism of the asymmetric addition and the epoxidation reaction.

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Supporting Information Available: Procedures and full characterization of new compounds (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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